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# Catechin Attenuates the Effect of Combined Arsenic and Deltamethrin Toxicity by Abrogation of Oxidative Stress and Inflammation in Wistar Rats

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**Abstract:** This study was aimed at evaluating the protective role of catechin (CT) against toxicity induced by combined exposure to arsenic (As) and deltamethrin (DM) in rats. Thirty-five (35) male Wistar rats were divided into 5 groups of 7 animals each. Treatment of each group was as follows: Control (C) administered corn oil (1ml kg<sup>-1</sup>), catechin only (CT) at 40mg kg<sup>-1</sup>, As+DM administered As (100ppm) in their drinking water and DM at a dose of 7.5mg kg<sup>-1</sup> (1/20<sup>th</sup> LD<sub>50</sub>), As+DM-CT<sub>40</sub> treated as As+DM in addition to oral administration of CT at 40mg kg<sup>-1</sup> while the last group, As+DM-CT<sub>80</sub> received the same treatment as As+DM, along with oral CT treatment at a dose of 80mg kg<sup>-1</sup>. The treatment lasted for 28 days. Effect of the treatment in inducing oxidative damage was appraised by estimating levels of lipid peroxidation, protein oxidation, glutathione and total antioxidant capacity in the liver, kidney, and testis of the rats. Also, the activities of superoxide dismutase, catalase, and glutathione peroxidase were assayed in the tissues. For the evaluation of inflammation, plasma levels of interleukin-6, tumor necrosis factor-alpha and 8-nitroguanine were determined. The result showed that the combination of As and DM gave rise to marked alterations of these parameters but supplementation with CT attenuated these effects.

**Keywords:** Arsenic, Deltamethrin, Oxidative Stress, Inflammation, Catechin

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## 1. Introduction

Humans are exposed to a large number of environmental contaminants. With the myriad of pollutants present in the environment, exposure to complex mixture of environmental toxic agents is the rule rather than the exception. Among these environmental agents are metals and pesticides, which are highly ubiquitous owing to their widespread use [1, 2]. One of the most hazardous metals according to the Agency for Toxic Substances and Disease Registry is arsenic [3]. Humans are exposed to this metalloid through contaminated water, soil and through occupational exposure [4]. Exposure to arsenic (As) has been associated with a wide range of toxic effects [5, 6]. These toxic effects include cancer of various

types, diabetes mellitus, hepatomegaly, cerebrovascular and cardiovascular diseases among many others [7]. Arsenic generally exerts its toxicity through the generation of reactive oxygen species and the disruption of cellular antioxidant defense system [8].

Another important group of environmental contaminant is the pesticides. The attempts to improve the quality and quantity of crop yields have resulted in a significant rise in the use of pesticides worldwide. Pesticides use has also increased in household activities. The uncontrolled use of these pesticides has led to serious health problems, which is of concern in public health [9]. According to some estimates, millions of people are suffering from severe pesticide poisoning with hundreds of thousands leading to fatalities [10]. Deltamethrin (DM), a type II pyrethroid is a pesticide

extensively used as an ectoparasiticide on animals and as an insecticide in agriculture and public health programs [11]. Its exposure has been linked to the generation of reactive oxygen species and the resultant oxidative stress in various tissues [12, 13].

Individual toxic effects of arsenic and deltamethrin have been well documented, and studies have highlighted the therapeutic potential of several natural agents in protecting against the toxicity induced by either of these toxicants [14-17]. There is however, a dearth of information on the protective effect of natural agents against multiple toxic exposure. This study therefore, embarked on an investigation of the efficacy of catechin, a polyphenol against toxicity induced by combined exposure to arsenic and deltamethrin in rats.

## 2. Materials and Methods

### 2.1. Chemicals

Sodium arsenite, catechin, xylol orange, tripyridyl-5-triazine (TPTZ), were procured from Sigma-Aldrich (Munich, Germany). Commercial grade deltamethrin marketed as Deltaforce® (Sobero Organics) was obtained from Irorun Agbe Agrochemical Company (Ogbomoso, Nigeria). ELISA kits for interleukin-6, TNF- $\alpha$ , and 8-nitroguanine determinations were by Abcam (UK), RayBiotech Inc, (USA), and Cell Biolabs Inc. (USA), respectively. Other chemicals and reagents were all of analytical grade.

### 2.2. Animals and Treatment

Thirty five male Wistar rats weighing 130-150g procured from the Animal House, Faculty of Basic Medical Sciences, Ladoko Akintola University of Technology were used for the study. The animals were kept in plastic cages under controlled laboratory conditions of normal light-dark cycle (12h light/dark) and temperature ( $25 \pm 2^\circ\text{C}$ ). The animals were fed with pelleted diet and water ad libitum. All animal experiments were executed according to the guidelines approved by the Research Ethical Committee of the Faculty of Basic Medical Sciences, Ladoko Akintola university of Technology, Nigeria.

### 2.3. Experimental Design

After a week acclimation period, the rats were weighed and randomly assigned into five groups of seven animals each. Group I (C) served as the control and was given corn oil only at a  $1\text{ml kg}^{-1}$ , while group II (CT) was given CT only at a dose of  $40\text{mg kg}^{-1}$ . Group III (As+DM) was administered with As (100ppm) in their drinking water (Ravuri, 2014) and DM by gavage at a dose of  $7.5\text{mg kg}^{-1}$  ( $1/20^{\text{th}}$  LD<sub>50</sub>). Group IV (As+DM-CT<sub>40</sub>) was treated as group III in addition to oral administration of CT at  $40\text{mg kg}^{-1}$ . Group V (As+DM-CT<sub>80</sub>) received the same treatment as group III along with oral CT treatment at a dose of  $80\text{mg kg}^{-1}$ . The different regimens were administered once daily for 28 days. The dose of one-twentieth LD<sub>50</sub> for DM was chosen to produce a sub-lethal

toxic effect in the animal, while those of CT were based on its therapeutic efficacy, as earlier reported [18]. At the end of the experiment, rats were lightly anaesthetized with ether and blood collected by cardiac puncture. Liver, kidney and testes were excised, weighed, and processed for biochemical assays.

### 2.4. Biochemical Analysis

#### 2.4.1. Protein Estimation

The protein content of the organs was determined by the method of Lowry et al [19], using bovine serum albumin as a standard.

#### 2.4.2. Estimation of Lipid Peroxidation Levels

Lipid peroxidation in tissues was estimated by the TBA reaction with malondialdehyde, a product of lipid peroxidation process [20]. Tissue homogenate prepared in 0.15M KCl was mixed thoroughly with a stock solution containing 15% w/v trichloroacetic acid, 0.375% w/v thiobarbituric acid and 0.25M HCl. The mixture was heated in a boiling water bath for 15min and cooled. The mixture was centrifuged at 1000g for 10min and the absorbance of the supernatant determined at 535nm. Using an extinction coefficient of  $1.56 \times 10^5\text{M}^{-1}\text{cm}^{-1}$ , the concentration of MDA was calculated and expressed as nmol per gram of wet tissue.

#### 2.4.3. Determination of Lipid Hydroperoxides Concentrations

Lipid hydroperoxides concentrations in plasma and tissue homogenates were determined using the method of Nourooz-Zadeh et al. [21]. Tissue homogenate was mixed with either 10mM TPP in methanol or with methanol and incubated for 30min at room temperature. FOX2 reagent was added and the mixture incubated for another 30min. The mixture was centrifuged to remove flocculated material at 12,000g for 10min after which the absorbance was read at 560nm.

#### 2.4.4. Estimation of Protein Oxidation

Advanced oxidized protein products (AOPP) was measured to assess the level of protein oxidation in the tissue [22]. Homogenate of the sample was prepared with phosphate buffered saline after which 1.16M potassium iodide and acetic acid were added. The absorbance of the reaction mixture was read at 340nm and an extinction coefficient of  $26\text{lmM}^{-1}\text{cm}^{-1}$  was used in calculating AOPP concentrations.

#### 2.4.5. Determination of Total Antioxidant Capacity (TAC)

The TAC in samples was estimated using the ferric reducing antioxidant power (FRAP) method described by Benzie and Strain [23]. The FRAP method measures the antioxidant potential through the reduction of ferric tripyridyl-5-triazine ( $\text{Fe}^{3+}$ -TPTZ) to its ferrous form ( $\text{Fe}^{2+}$ ). Briefly, FRAP working reagent (300mM acetate buffer, pH 3.6, 10mM 2,4,6-tripyridyl-S-triazine in 40mM HCl and 20mM  $\text{FeCl}_3$  in ratio 10:1:1) was pre-warmed and then mixed with 10% homogenate of the sample. The procedure was performed at  $37^\circ\text{C}$ , and the absorbance was read at

593nm.

#### 2.4.6. Determination of Reduced Glutathione (GSH)

Reduced glutathione content in the organs was determined using the method of Moron *et al.* [24]. Tissue homogenate was treated with 0.5ml Ellman's reagent in 0.1% sodium citrate. After that, phosphate buffer and DTNB were added before the absorbance was read at 412nm.

#### 2.4.7. Determination of Catalase (CAT) Activity

The activity of CAT was measured using the method of Aebi [25]. The assay mixture consisting of supernatant tissue homogenate (0.1ml), 50mM phosphate buffer (1.9ml) and 30mM H<sub>2</sub>O<sub>2</sub> (1ml) was maintained at 20°C. The decomposition of H<sub>2</sub>O<sub>2</sub> was monitored spectrophotometrically at 240nm. The change in absorbance was the measure of CAT activity and expressed as nmol/mg protein.

#### 2.4.8. Determination of Superoxide Dismutase (SOD) Activity

The method of Misra and Fridovich [26] was used in determining SOD activity in the tissues. The addition of 0.01% epinephrine (0.3ml) to the mixture containing 2.5ml carbonate buffer (0.05M) and 0.2ml sample initiated the reaction. Change in absorbance was measured at 480nm, and the activity was expressed as unit per milligram of protein.

#### 2.4.9. Determination of Glutathione Peroxidase (GPx) Activity

GPx activity was determined using H<sub>2</sub>O<sub>2</sub> as a substrate in the presence of reduced glutathione [27]. The reaction was initiated by adding 0.2mM H<sub>2</sub>O<sub>2</sub> to the mixture containing 0.2ml of phosphate buffer (0.4M), 0.1ml of sodium azide (10mM), 0.2ml of tissue homogenate and 0.2ml of GSH. The GSH content was quantified using Ellman's reagent. The activity was expressed as unit per milligram protein where a unit is mmol of GSH consumed per minute.

#### 2.4.10. Determination of Plasma Interleukin-6 (IL-6), Tumor Necrosis Factor-alpha (TNF-α) and Nitritive Nucleic Acid Damage

Plasma concentrations of IL-6 and TNF-α were determined using commercial rat ELISA kits (Abcam, UK and Ray Biotech Inc, USA, respectively), following the instructions in respective kit manuals. The absorbance was monitored at 450nm on a Hawksley HA-1600 microplate reader (Hawksley, London, U.K.). Plasma concentration of 8-nitroguanine (8-NO<sub>2</sub>-Gua) was evaluated using the nitrosative DNA/RNA damage ELISA kit (Cell Biolabs, Inc. USA), following the procedure provided by the manufacturer.

#### 2.5. Statistical Analysis

Data were analyzed by one-way analysis of variance, followed by Tukey's multiple comparisons test. Results were presented as mean ± standard deviation (SD) and values were considered statistically significant at p < 0.05. Data were analyzed using GraphPad Prism for Windows, version 6.01 (GraphPad Software, Inc., San Diego, CA, USA.).

### 3. Results

#### 3.1. Effect on Lipid Peroxidation and Protein Oxidation Indices

Table 1 shows the combined effects of arsenic and deltamethrin, along with catechin treatment on MDA, LOOH and AOPP levels in liver, kidney and testis tissues. Administration of CT to normal rats exerted no significant difference on the lipid peroxidation and protein oxidation markers. Tissue MDA, LOOH, and AOPP contents were however, increased significantly on exposure to As in combination with DM. CT treatments caused significantly reduction of these parameters in the co-exposed animals. The polyphenol particularly restored hepatic MDA and AOPP levels to normal, while drastically lowering all the markers of oxidative injury in all other organs.

Table 1. Effects of catechin treatment on MDA, LOOH and AOPP levels in liver, kidney, and testis of rats co-exposed to arsenic and deltamethrin.

Groups	Liver			Kidney
	MDA (nmol/g tissue)	LOOH (nmol/g tissue)	AOPP (μmol/g tissue)	MDA (nmol/g tissue)
Control	31.29 ± 3.07 <sup>a</sup>	7.14 ± 0.63 <sup>a</sup>	39.27 ± 1.95 <sup>a</sup>	18.58 ± 1.24 <sup>a</sup>
CT	30.86 ± 2.72 <sup>a</sup>	7.21 ± 0.43 <sup>a</sup>	37.16 ± 1.72 <sup>a</sup>	18.10 ± 1.11 <sup>a</sup>
As+DM	53.65 ± 3.81 <sup>b</sup>	21.05 ± 1.54 <sup>b</sup>	48.07 ± 2.34 <sup>b</sup>	34.25 ± 2.34 <sup>b</sup>
As+DM-CT <sub>40</sub>	35.12 ± 2.35 <sup>a</sup>	11.36 ± 1.89 <sup>c</sup>	36.94 ± 2.13 <sup>a</sup>	26.76 ± 1.92 <sup>c</sup>
As+DM-CT <sub>80</sub>	32.33 ± 2.84 <sup>a</sup>	9.05 ± 0.61 <sup>d</sup>	36.10 ± 2.73 <sup>a</sup>	23.55 ± 1.81 <sup>d</sup>

Table 1. Continued.

Groups	Kidney		Testis		
	LOOH (nmol/g tissue)	AOPP (μmol/g tissue)	MDA (nmol/g tissue)	LOOH (nmol/g tissue)	AOPP (μmol/g tissue)
Control	5.01 ± 0.34 <sup>a</sup>	12.61 ± 0.32 <sup>a</sup>	11.55 ± 1.33 <sup>a</sup>	1.85 ± 0.18 <sup>a</sup>	40.67 ± 1.69 <sup>a</sup>
CT	4.76 ± 0.31 <sup>a</sup>	12.79 ± 0.28 <sup>a</sup>	10.12 ± 0.84 <sup>a</sup>	1.56 ± 0.21 <sup>a</sup>	38.31 ± 2.05 <sup>a</sup>
As+DM	12.52 ± 1.57 <sup>b</sup>	41.61 ± 4.65 <sup>b</sup>	28.24 ± 2.37 <sup>b</sup>	4.89 ± 0.52 <sup>b</sup>	77.50 ± 5.12 <sup>b</sup>
As+DM-CT <sub>40</sub>	7.74 ± 0.85 <sup>c</sup>	13.66 ± 1.33 <sup>ac</sup>	15.56 ± 1.84 <sup>c</sup>	2.62 ± 0.25 <sup>c</sup>	37.50 ± 2.00 <sup>a</sup>
As+DM-CT <sub>80</sub>	7.18 ± 0.49 <sup>c</sup>	16.60 ± 1.28 <sup>c</sup>	13.56 ± 1.55 <sup>ac</sup>	2.17 ± 0.24 <sup>ac</sup>	37.57 ± 1.33 <sup>a</sup>

MDA: malondialdehyde; LOOH: lipid hydroperoxide; AOPP: advanced oxidized protein product.

Data are expressed as means ± S. D. of seven animals per group. Values in the same column, not sharing the same superscript are significantly different from each other at p < 0.05.

### 3.2. Effects on Antioxidant Systems

Table 2 depicts the enzymatic and non-enzymatic antioxidant status in control rats and groups exposed to combined As and DM alone or co-administered with CT. There were no significant changes in activities of SOD, CAT, and GPx in rats treated with CT alone. Combined As and DM treatment caused significant ( $p < 0.05$ ) reduction in the activities of SOD and GPx in the liver (65% and 40%, respectively), kidney (70% and 40%, respectively), and testis (63% and 36%, respectively) when compared with the control. However, co-treatment with CT reversed this trend in all the organs, with CT administration fully restoring hepatic SOD activity and GPx activities in both the liver and kidney. Combined As and DM also reduced hepatic and renal CAT

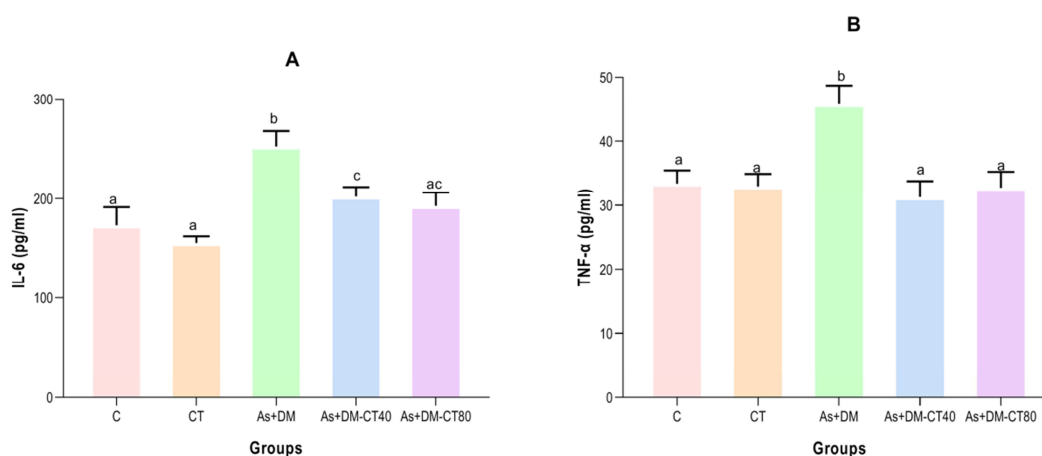
activities but the enzyme activity was increased in the testis. CT administration at high dose restored the enzyme activities in both the kidney and testis in the exposed rats. Hepatic CAT activity was significantly increased by both CT doses, when compared with the animals treated with combined As and DM alone. The levels of GSH and TAC were significantly ( $p < 0.05$ ) decreased in the liver, kidney and testis of As and DM treated rats when compared with the control. No significant change was observed in the hepatic content of GSH of animals co-exposed to As and DM but treated with either doses of CT. The kidney and testis GSH levels, along with TAC in all the tissues of rats exposed to the chemical mixture but treated with CT were significantly increased above that of the animals treated with combined As and DM alone.

**Table 2.** Effects of epicatechin treatment on enzymatic and non-enzymatic antioxidant status in liver and kidney of Cypermethrin intoxicated rats after 14 days.

Groups	SOD (U/mg protein)	CAT (nmol/mg protein)	GPx (U/mg protein)	GSH ( $\mu\text{mol/g}$ tissue)	TAC ( $\mu\text{mol/g}$ tissue)
<b>Liver</b>					
Control	12.45 $\pm$ 0.92 <sup>a</sup>	31.22 $\pm$ 2.46 <sup>a</sup>	91.47 $\pm$ 4.85 <sup>a</sup>	20.84 $\pm$ 3.13 <sup>a</sup>	99.68 $\pm$ 6.90 <sup>a</sup>
CT	12.14 $\pm$ 0.94 <sup>a</sup>	28.66 $\pm$ 3.14 <sup>ac</sup>	93.92 $\pm$ 5.77 <sup>a</sup>	23.63 $\pm$ 2.34 <sup>a</sup>	142.20 $\pm$ 17.07 <sup>b</sup>
As+DM	4.42 $\pm$ 0.51 <sup>b</sup>	17.58 $\pm$ 2.02 <sup>b</sup>	54.67 $\pm$ 3.21 <sup>b</sup>	11.54 $\pm$ 1.12 <sup>b</sup>	70.20 $\pm$ 8.78 <sup>c</sup>
As+DM-CT <sub>40</sub>	8.75 $\pm$ 0.93 <sup>c</sup>	26.23 $\pm$ 1.10 <sup>c</sup>	86.25 $\pm$ 8.39 <sup>a</sup>	18.74 $\pm$ 1.27 <sup>a</sup>	199.70 $\pm$ 13.32 <sup>d</sup>
As+DM-CT <sub>80</sub>	11.83 $\pm$ 1.12 <sup>a</sup>	25.47 $\pm$ 2.27 <sup>c</sup>	92.38 $\pm$ 7.68 <sup>a</sup>	20.24 $\pm$ 1.70 <sup>a</sup>	211.80 $\pm$ 12.24 <sup>d</sup>
<b>Kidney</b>					
Control	7.23 $\pm$ 0.55 <sup>a</sup>	12.19 $\pm$ 0.80 <sup>a</sup>	45.17 $\pm$ 3.68 <sup>a</sup>	13.88 $\pm$ 0.95 <sup>a</sup>	219.60 $\pm$ 9.73 <sup>a</sup>
CT	6.86 $\pm$ 0.57 <sup>ac</sup>	11.84 $\pm$ 1.02 <sup>a</sup>	47.24 $\pm$ 2.95 <sup>a</sup>	14.27 $\pm$ 1.04 <sup>a</sup>	222.10 $\pm$ 11.63 <sup>ad</sup>
As+DM	2.20 $\pm$ 0.34 <sup>b</sup>	4.83 $\pm$ 0.37 <sup>b</sup>	27.07 $\pm$ 3.13 <sup>b</sup>	5.57 $\pm$ 0.38 <sup>b</sup>	179.70 $\pm$ 14.05 <sup>b</sup>
As+DM-CT <sub>40</sub>	5.86 $\pm$ 0.89 <sup>c</sup>	10.26 $\pm$ 1.28 <sup>c</sup>	31.83 $\pm$ 3.33 <sup>b</sup>	10.61 $\pm$ 1.11 <sup>c</sup>	351.60 $\pm$ 12.66 <sup>c</sup>
As+DM-CT <sub>80</sub>	6.11 $\pm$ 0.47 <sup>c</sup>	11.18 $\pm$ 1.00 <sup>ac</sup>	41.83 $\pm$ 3.79 <sup>a</sup>	12.30 $\pm$ 0.87 <sup>c</sup>	239.00 $\pm$ 12.00 <sup>c</sup>
<b>Testis</b>					
Control	4.94 $\pm$ 0.31 <sup>a</sup>	3.32 $\pm$ 0.30 <sup>a</sup>	16.42 $\pm$ 1.22 <sup>a</sup>	4.97 $\pm$ 0.37 <sup>a</sup>	29.53 $\pm$ 1.44 <sup>a</sup>
CT	5.21 $\pm$ 0.44 <sup>a</sup>	3.43 $\pm$ 0.49 <sup>a</sup>	17.75 $\pm$ 2.01 <sup>a</sup>	4.64 $\pm$ 0.31 <sup>ac</sup>	32.94 $\pm$ 1.14 <sup>b</sup>
As+DM	1.82 $\pm$ 0.70 <sup>b</sup>	5.87 $\pm$ 0.68 <sup>b</sup>	10.49 $\pm$ 0.94 <sup>b</sup>	2.11 $\pm$ 0.15 <sup>b</sup>	13.02 $\pm$ 1.48 <sup>c</sup>
As+DM-CT <sub>40</sub>	3.95 $\pm$ 0.21 <sup>c</sup>	4.85 $\pm$ 0.43 <sup>c</sup>	14.89 $\pm$ 1.12 <sup>c</sup>	3.81 $\pm$ 0.30 <sup>cd</sup>	22.95 $\pm$ 2.34 <sup>d</sup>
As+DM-CT <sub>80</sub>	4.26 $\pm$ 0.30 <sup>c</sup>	3.85 $\pm$ 0.31 <sup>a</sup>	15.18 $\pm$ 1.34 <sup>c</sup>	4.32 $\pm$ 0.35 <sup>d</sup>	26.13 $\pm$ 2.90 <sup>e</sup>

SOD: superoxide dismutase; CAT: catalase; GPx: glutathione peroxidase; GSH: reduced glutathione; TAC: total antioxidant capacity. Data are expressed as means  $\pm$  S. D. of seven animals per group. Values in the same column for a compartment, not sharing the same superscript are significantly different from each other at  $p < 0.05$ .

### 3.3. Effect on Pro-inflammatory Cytokines



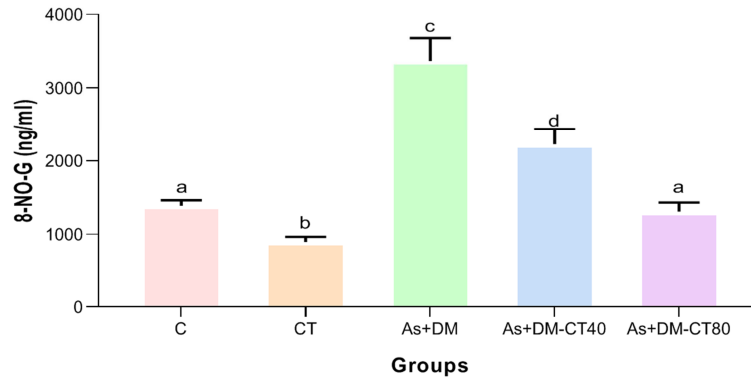
**Figure 1.** Effect of catechin treatment on plasma IL-6 (A) and TNF- $\alpha$  (B) levels in rats co-exposed to As and DM.

The concentrations of IL-6 and TNF- $\alpha$  in the plasma of combined As and DM-exposed and CT-administered rats are

shown in Figure 1. Administration of CT alone has no significant effect on the levels of inflammatory markers.

However, the pro-inflammatory cytokines levels were increased significantly ( $p < 0.05$ ) in the co-exposed rats. The concentrations of IL-6 and TNF- $\alpha$  in the plasma were increased by 46% and 37% respectively above the control.

### 3.4. Assessment of Nitrosative Damage to Nucleic Acids



**Figure 2.** Effect of catechin treatment on plasma 8-NO<sub>2</sub>-Gua levels in rats co-exposed to As and DM.

Figure 2 depicts the levels of nitrosative damage in the rats. Co-exposure to As and DM is characterized by a significant increase in plasma concentration of 8-NO<sub>2</sub>-Gua in the rats ( $p < 0.05$ ). Combined As and DM intoxication resulted in a 142% increase in the level of 8-NO<sub>2</sub>-Gua when compared with the control. Co-treatment of As and DM-exposed rats with CT significantly reduced the formation of nitrosative nucleic acids. The nitrosative nucleic acid content in the rats treated with high dose CT was not significantly different from the control. Interestingly, CT alone reduced 8-NO<sub>2</sub>-Gua level in normal rat by 36%.

## 4. Discussion

Environmental pollutants such as pesticides and heavy metals are known to cause serious health effects through the induction of wide range of toxicological events and biochemical dysfunctions. Some earlier studies on combined exposure to pesticides and heavy metals have reported among others, immunotoxic, hematologic, and calcium homeostasis disruptive effects of these environmental contaminants [28, 29]. There is however, a paucity of information on the use of natural agents in preventing or ameliorating these toxic effects. The present study therefore, describes the effectiveness of oral CT administration in preventing oxidative stress and inflammation induced by combined As and DM exposure in the rat. Oxidative imbalance induced by arsenic and DM has been well reported, resulting from increased free radical generation and depletion of antioxidant defense system induced by these toxicants [12, 30]. In the present study, results show that the combination of the toxicants caused an imbalance in the oxidative status and increased inflammation in the exposed rats which CT was able to attenuate.

Even though liver and kidney play roles in detoxification, their chronic exposure to xenobiotics and their derivatives make them susceptible to oxidative damage [31]. On the other hand, the high polyunsaturated fatty acid content along

CT co-treatment significantly reduced the elevated pro-inflammatory cytokines induced by the combination of As and DM, with the high CT dose treatment restoring the levels to normal.

with the low antioxidant capacity make testicular tissues prone to attack from ROS [32]. In this study, co-exposure to As and DM resulted in the inhibition of all the antioxidant enzyme activities measured except in the testis, where the toxicants caused an increase in CAT activity. Inhibition of antioxidant enzyme activity has been reported following exposure of animals to either As or DM [33-36]. The observed decrease in SOD, CAT and GPx activities may be due to oxidative inactivation of the proteins arising through the damaging effect of free radicals generated by both As and DM. The damage may result from the accumulation of the insecticide in the tissue, together with the disruption of the protein's thiol groups by arsenic [31]. Inhibition of these enzymes is often accompanied by a depletion of GSH as evidenced in this study, and/or increased hydrogen peroxide generation, giving rise to oxidative stress [37]. The increase observed in testicular CAT activity may be an adaptive response by the tissue to counter the increased generation of ROS through increased expression of the protein. Pretreatment with CT however, ameliorated the disruption caused by the co-exposure to As and DM to the antioxidant enzyme activities, with a total restoration of activity generally observed with the high dose of the polyphenol.

GSH and GPx are detoxifying agents that convert H<sub>2</sub>O<sub>2</sub> and lipid peroxides to nontoxic products. The depleted GSH concentration caused by increased oxidative stress burden could induce lipid and protein oxidation [38]. In our study, this is reflected by the marked increase in tissue MDA, LOOH, and AOPP contents in the co-exposed rats. Several authors have also demonstrated increased oxidative stress following co-exposure to multiple chemical mixtures [11, 39, 40]. Either As or DM alone, can enhance lipid peroxidation through interaction with cellular membranes [41, 42]. Increased generation of lipid peroxides can result in the depletion of lipid soluble antioxidant system like GSH, leading to the inhibition of GSH-dependent antioxidant enzymes, such as GPx. The resultant effect of this would be

an increased susceptibility of the enzymatic antioxidant system to oxidative damage, contributing to their reduced activity. The increase in ROS production could also contribute to the generation of oxidized proteins observed in all the tissues investigated, the resulting modification might have added to the decrease in enzyme activities found in this study. CT treatment significantly attenuated the lipid peroxidation and reversed the oxidant-mediated protein damaging effects of the toxicants. Co-exposure to As and DM was characterized by reduced hepatic, renal and testicular antioxidant potential as indicated by the lowered FRAP level in the rats. The relative cumulative capacity of the total antioxidant system was reflected by the antioxidant capacity estimation. The reduction in TAC by AS and DM co-administration is an indication of the depletion of the antioxidant system capacity in the organs. Results from this study, however, suggested that concomitant treatment with CT normalized the TAC level in all the organs.

The toxic effects elicited by co-exposure to As and DM were all significantly reduced or completely reversed by treatment with CT. Although reports are not readily available about the protective effect of CT against combined As and DM induced toxicity, there are earlier reports of its ability to protect against chemically induced oxidative stress and inflammation [43, 44]. In the present study, CT restored the activities of the antioxidant enzymes SOD, CAT and GPx, while suppressing the generation of MDA, LOOH, and AOPP in the organs. The improvement of the oxidative mediated impairment of the tissues in As and DM co-exposed rats by CT indicated that the polyphenol effectively scavenged free radicals and attenuated the oxidative damage. Previous studies have shown CT to be an effective agent able to scavenge and remove deleterious toxic radicals that could disrupt antioxidant enzyme activities and cause tissue injury [43, 45]. Studies have also demonstrated that catechin could inhibit excessive oxidative stress through direct or indirect antioxidant effects and can reduce oxidative damage to tissues through the promotion of antioxidant substances such as GPx and GSH [44]. The antioxidant properties of CT may thus be employed to improve the redox status in the tissues following exposure to multiple toxicants, such as As and DM.

The present investigation also revealed that co-administration of As and DM triggered significant increase in the level of pro-inflammatory cytokines, IL-6 and TNF- $\alpha$ , as well as, the nitrative nucleic acid marker, 8-NO<sub>2</sub>-Gua. Increased oxidative stress can induce inflammation through the activation of redox-sensitive factors [46]. The elevated IL-6 and TNF- $\alpha$  levels observed could thus, be a result of these redox-sensitive factors, activated by combined As and DM induced oxidative stress. Several studies have shown that either toxicant can induce the expression of proinflammatory cytokines [7, 47]. 8-NO<sub>2</sub>-Gua is a product of the interaction of the highly reactive peroxynitrite, formed when nitric oxide synthase reacts with superoxide, and guanine, during inflammation [48]. Elevated plasma 8-NO<sub>2</sub>-Gua level in the rats further corroborates the inducement of inflammation by co-administration of As and DM. The effect

of CT on 8-NO<sub>2</sub>-Gua level in these rats reflected the anti-inflammatory property of CT against the inflammatory response induced by co-exposure to As and DM. Plasma TNF- $\alpha$ , IL-6 and 8-NO<sub>2</sub>-Gua levels were normalized by CT administration. Catechins are reported to exert their anti-inflammatory properties by regulating the activation of inflammation-related oxidative stress-related cell signaling pathway, such as nuclear factor-kappa B (NF- $\kappa$ B), mitogen activated protein kinases (MAPKs), transcription factor nuclear factor (erythroid-derived 2)-like 2 (Nrf2), signal transducer and the activator of transcription 1/3 (STAT1/3) pathways. Catechin also exerts its anti-inflammatory function by decreasing the expression of inflammatory cytokines [43].

## 5. Conclusion

In conclusion, the present study indicate that combined exposure to As and DM resulted in marked imbalance in oxidative status and aggravated inflammatory response in the tissues of rats. These alterations were, however, mitigated by treatment with CT. The study, therefore, showed that supplementation with CT has beneficial effect on toxicities evoked by co-exposure to As and DT, and potentially, on injuries caused by exposure to other multiple toxicants.

## Statement of Competing Interests

The authors have no competing interests.

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