Quality Assessment of Oil Used for Frying of Potato Chips: In the Case of Wolkite University Community

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Abstract: Frying is an old and widely used method of cooking and processing food. Typically, a food is immersed in heated oil for a short duration in a process known as immersion-oil frying. Consumers tend to associate color with flavor, safety, storage time, nutrition, and level of satisfaction, due to the fact that color correlates well with physical, chemical, and sensorial evaluations of food quality. Indicators of poor oil quality include elevated FFA, low smoke point, change of color, low iodine value, peroxide value, total polar material, high foaming properties and increased viscosity. Deep frying and the use of recycled oil for many times is a commonly practiced in commercial and sometimes in domestic cooking processes. The way of this cooking method may create a lipid peroxidation product which brings harm to human health. Quality assessment of oil used for frying of potato chips in the case of Wolkite university community studied. The result of physicochemical properties of oil were investigated from sample selected randomly around Wolkite town and these are free fatty acid, peroxide value, smoking point and iodine value of recycled oils was higher than the recommended values. The strength of current evidence makes it reasonable to recommend complete avoidance of fried foods or at most infrequent to moderate fried food consumption within the context of an overall healthy dietary pattern around Wolkite town. Generally, the present study shows the recycled oil used were not as standards of FAO and also the people of fried food makers were not having awareness about the impacts of recycled oils.

Keywords: Fat, Oil, Recycled and Frying

1. Introduction

Frying is an old and widely used method of cooking and processing food. Typically, a food is immersed in heated oil for a short duration in a process known as immersion-oil frying. Numerous type of edible oils of plant and animal origin are used in frying. During the last five decades, the Western food industry has become increasingly dependent on the frying process to manufacture a variety of snack foods. Fried foods such as potato chips, French fries, and fried fish and chicken have gained worldwide popularity. A wide variety of products are fried on an industrial scale. Among the most popular are potato chips (also called potato crisps), expanded snacks, roasted nuts, French fries (also called potato chips), extruded noodles, doughnuts, and frozen foods covered with batter, such as fish & chicken. In the potato chip industry, the quality of each batch of potato tubers must be tested before processing, and of the various quality criteria the visual aspect, especially the color, is of great importance [1]. Color of potato chips is the first quality parameter evaluated by consumers, and is critical in the acceptance of the product. Consumers tend to associate color with flavor, safety, storage time, nutrition, and level of satisfaction, due to the fact that color correlates well with physical, chemical, and sensorial evaluations of food quality. Indicators of poor oil quality include elevated FFA, low smoke point, change of color, low iodine value, peroxide value, total polar material, high foaming properties and increased viscosity [2]. Restaurants and other food services had monitored the changes in physical properties of frying oils which used as an indicator of oil quality. For example, sample of frying oil could be discarded if the oil presented in dark, too much smoke, strong odor, greased texture, or when a persistent foam layer of specified thickness is observed. Nevertheless, before an operator found the physical effects, the oil has typically undergone considerable decomposition [3].
Oxidation is a major deteriorating process that gives huge implication in the conditions of the quality and assessment such as developing the rancidity and off-flavor of fats and oils [4]. Basically, frying oils undergo common degradation and complex chemical alteration when heated. Acceleration of the deterioration in frying oil and also producing a number of polar molecules as resulted by viscosity and density data of oils can be caused by the presence of air and water. Hydroperoxide is from the formation of primary oxidation products that develop from triacylglycerol, which may later decompose in order to become lower molecular weight compounds like free fatty acids, alcohols, aldehydes, and ketones which ultimately leading to a rancid product[5]. In addition, the oxidation process that occurs in oils, whether in storage or heating, condition may contribute in low nutritional quality, alter the flavor and may generate toxic compound [6]. Indeed, [7] claimed that, higher temperature use during frying can make the oil form a radical compound that give bad impact to health. Therefore, the peroxide compound in deep-frying cooking oil has been measured. Acrylamide, which is formed in potatoes during frying and is highly related to the color of potato chips, is suspected to have critical implications for human health, since it has recently been found to be a carcinogen in rats [8]. Potato chip color is affected by the Maillard reaction, which depends on the content of reducing sugars, amino acids, or proteins at the surface. Generally, potato tubers that contain more than 2 percent of reducing sugars are discarded for frying, since they generate too dark a coloration. In general, foods rich in carbohydrate and prepared at temperature above 160°C exhibit acrylamide formation and this may be related to the degree of browning. Browning is a result of a series of reactions known as Maillard reaction. Maillard reaction, apart from providing desirable color, contributes to the flavor and aroma of the final product. Studies conducted so far indicated that foods with high carbohydrate content such as potatoes, develop high levels of acrylamide when heated at high temperature. It has been suggested that the amino acid asparagine and reducing sugars in Maillard reaction accounted for the majority of acrylamide in potato products.

1.1. Statement of the Problem

Deep frying and the use of recycled oil for many times is a commonly practiced in commercial and sometimes in domestic cooking processes. The way of this cooking method may create a lipid per oxidation product which brings harm to human health. Most of the products are non-volatile, so they will be remained in the food or oils and give effect to the physical properties and form other radicals. This is due to the high temperatures in the existence of air and moisture, which make the occurrence of oxidative degradation of amino acids and the partial conversion of these lipids to volatile chain-scission products, non-volatile oxidized derivatives and polymeric or cyclic substances leading to the formation of toxic or carcinogenic compounds [9]. Therefore, this action research gives an overview of some commonly consumed fried products (potato chips) in Wolkite University, the effects of thermal oxidation on quality of oil used and possible health implications.

1.2. General Objective

To assess the quality of recycled oil used for frying of potato chips in the case of Wolkite University community.

1.2.1. Specific Objectives

1. To study chemical quality of recycled oil used of frying
2. To assess awareness of Wolkite university community in relation to the quality of oils used for frying

1.2.2. Significance of the Study

This particular action research had a great role on alleviating health problems related to fried food consumption specially potato chips by creating awareness to Wolkite town. As a result, the society consuming potato chip will be changed and take consideration of the status of the oil used for frying purpose. Since the oil quality has a considerable effect on the human health, acceptability and nutritional value of potato chips. In addition, the findings of this action research will give direction for further researches.

2. Literature Review

2.1. Application of Oils for Frying

The use of oil or fat for frying of food around the world, as confirmed by the increasing number of fast food restaurants and at home. Also, edible oil is a source of energy and essential fatty acids (FAs) and serves as a carrier of fat-soluble vitamins [10]. Fried foods are popular because of their rapid and easy production. Furthermore, fried foods are highly appreciated by consumers because owing to their appealing sensory properties such as color, flavor, texture, and palatability [11]. Regular consumption of deep-fried foods such as french-fries, fried chicken, and doughnuts has recently increased with economic development and a change in eating habits. Deep frying involves simultaneous heat and mass transfers in the food processing operation by immersing the food as a whole into the hot oil at temperatures of 180°C or higher [12]. However, during the frying process, oil or fat is often recycled for several batches, allowing moisture and air to be mixed into the hot oil. As a result, these fats and oils undergo thermal and oxidative decomposition, and polymers formed under these conditions are harmful to health [13]. The associated reactions increase the viscosity of the oils, darken their color, increase the foaming, and decrease the smoke point. Overall, deep-frying deteriorates the quality of the oil. The choice of frying oil or fat depends on many factors such as availability, price, frying performance, flavor, and stability of the product during storage. The most popular oil used for frying products is palm oil (PA) or its fraction, followed by soybean oil (SB), canola oil (CA), and lard (LA). Hydrogenated vegetable oils such as sunflower and rapeseed oil are also [14]. Abused frying oils may contain harmful concentrations of degradation products, and so, frying oil quality is also a very important issue concerning food safety and consumers’ health [15].
2.2. Quality of Cooking Oil

Edible oil is one of the chosen ingredients of the diet that used for cooking purposes. Researchers had studied the effect of temperature on the stability, viscosity, peroxide value, and iodine value to assess the quality and functionality of the oil [16]. It is necessary to observe the quality of oil in order to avoid the use of abused oil, bad health consequences due to high consuming foods fried in degraded oil, to keep the quality of fried foods and to minimize the production costs associated with early disposal of the frying medium. Oil that rich with unsaturated fatty acids may allowed in rapid degradation because of the oxidation activity [17]. Atmospheric oxygen responds instantly with lipid and other organic compounds of the oil to cause structural degradation in the oil, which leads to loss of quality of food and is harmful to human health (Bhattacharya et al., 2008). Besides, the occurrence of hydrolysis is generally caused by the present of water that came from the fried food [18]. Then, increasing of free fatty acid value is mainly from the degradation of secondary oxidation products formed during heating and the presence of oxygen. The existence of moisture that may be provided by air could have an influence on increasing acidity. However, the drop off in the acidity level after 24 h and 30 h may be caused by the evaporation of the volatile free fatty acids under the heating effect. The polymerization of fatty acids during the heating process may also block the acid functions that are no longer assayed [19]. A various type of physical and chemical parameters of edible oil was used to examine the compositional quality of oils [20]. These physicochemical parameters include iodine value (IV), saponification value (SV), viscosity, density and peroxide value (PV) [21] clarified about hydrolytic rancidity that happen due to hydrolysis of fats and oils and it has potential in yielding free fatty acid (FFA). The level of acidity in 1g of sample can be determined by neutralize the FFA using milligrams of potassium hydroxide. A part from that, Europe is the only state in the world that has guidelines on frying oil safety [22]. Furthermore, the color change in oils is mainly caused by an accumulation of highly conjugated oxidation products. Color was closely correlated with all the assessed parameters. However, it needs to be considered with color measurement because they could be influenced by pigments extracted from the food [23].

2.3. Parameter in Measuring the Oil Quality

2.3.1. Viscosity and Density Measurement

A quality of fried food is also corresponding with the using oil because as the oil degrades, it will be changing the quality of the food products. Now a day, there are many physical test and chemical test found in the food industry. One of the methods and instrument that available to check on the oil quality is by measuring the viscosity. The designed instrument was to measure the formation of polymerized and oxidized matter (POM) contained in used cooking oil [23]. The viscosity of oils refers to the present of complex triglyceride (TGs) that is naturally present in oils. The alteration of fatty acids arrangement on the glycerol backbone of the triglyceride molecule can change the viscosity of the oils. Consequently, the chain length with the chemical properties of saturated and unsaturated chain give affects to the viscosity of oils [24]. Moreover, the viscosity and density decrease with an increase of unsaturation chain [25]. Sheer stress and temperature are also related to viscosity. However, sheer stress does not give much effect toward storage of oils which are used for edible purposes compared to the effect of temperature. When the temperature goes high the kinetic energy will be also increase to enhance the movement of the molecules and lowered the intermolecular forces. The layers of the liquid easily pass over one another and thus contribute to the reduction of viscosity. This nature is also revealed by another study that discussed the viscosity is depends on the molecular structure and decreases with the unsaturation of fatty acids [26]. The density of oils will be decreased when the temperature is arisen as well as when using the same oil for frying three times. The densities of oils were related to the standard range of 0.898–0.907 g/ml approved by the Standard Organization of Nigeria (SON, 2000). Frying process could undergo thermo-oxidative or lipid oxidation and hydrolytic reactions which results in deterioration in quality of the oil [27].

2.3.2. Saponification Value Measurement

Saponification value (SV) is an index of an average molecular mass of fatty acid in the oil sample. The lower value of saponification represent that the mean molecular weight of fatty acids is lower or that the number of ester bonds is less. This might imply that the fat molecules did not interact with each other.

2.3.3. Iodine Value Measurement

Iodine value (IV) is a measure the degree of unsaturation in a fat or vegetable oil. It verifying the oxidative stability of oils and can qualitatively determine the whole unsaturation of fats [29]. The result of low iodine values can contribute to its greater oxidative storage stability [28].

2.3.4. Free Fatty Acid Measurement

There are several methods to investigate the quality of oil is such as by measuring the free fatty acid (FFA) and total polar compound level [29]. FFA levels represent the mechanism of triacylglycerol elements had undergone hydrolytic degradation. Generally during the refining process, high level of FFAs compounds could be removed from the crude oils. Stated that, the refined oil has to be 0.1% less of FFA content. When the content of FFAs in frying oil is more than 1%, it is suggested to be discarded. Furthermore, the oils could be discarded if the FFA content had reached to 2% as claimed by the United States Department of Agriculture. Other than that, the limit of FFA in frying oil that have been proposed by some of the European countries as their regulatory rule ranging from 1.0 to 2.5%.

2.3.5. Peroxide Value Measurement

Peroxide value (PV) is related in measuring rancidity
reactions and can be used as an indication of the quality and stability of fats and oils. The peroxide value could increase corresponds with the storage time, contact with air of the oil samples and temperature [30]

3. Materials and Methods

The experiment of oil quality determination was conducted at Wolkite University in Food Process Engineering laboratory. And survey assessment was conducted at Wolkite University community.

3.1. Experimental Materials and SAMPLING

Oil samples were acquired from Wolkite town fried products makers. The acquired samples were selected based on random sampling techniques.

3.2. Methods

Percentage of free fatty acids, peroxide value, smoking point and iodine value were determined using the [31] in triplicate sample.

4. Results and Discussions

4.1. Dietary Quality Survey on Oil Fried Food Product Consumption

This Summary Report describes the methods used and the main findings with regard to fried food consumption, types of oil used, and health impacts of the oil recycled for frying. The survey was carried out by the Wolkite University community.

<table>
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<tr>
<th>No.</th>
<th>Question Type</th>
<th>Frequency</th>
<th>Percentage</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Do you know saturated and unsaturated oils?</td>
<td>42</td>
<td>100</td>
</tr>
<tr>
<td>2</td>
<td>Yes</td>
<td>28</td>
<td>66.67</td>
</tr>
<tr>
<td>3</td>
<td>No</td>
<td>14</td>
<td>33.33</td>
</tr>
<tr>
<td>4</td>
<td>Do you know how they can be prepared/processed?</td>
<td>42</td>
<td>100</td>
</tr>
<tr>
<td>5</td>
<td>Yes</td>
<td>42</td>
<td>100</td>
</tr>
<tr>
<td>6</td>
<td>No</td>
<td>0</td>
<td>0</td>
</tr>
<tr>
<td>7</td>
<td>Do you agree all fried products are produce under a sanitary condition in Wolkite town?</td>
<td>42</td>
<td>100</td>
</tr>
<tr>
<td>8</td>
<td>Yes</td>
<td>12</td>
<td>28.57</td>
</tr>
<tr>
<td>9</td>
<td>No</td>
<td>30</td>
<td>71.43</td>
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The study population consisted of 42 respondents of which 66.67% (28) knows saturated and unsaturated oils and 33.33% (14) haven’t know how about saturated and unsaturated oils. So, the type of oil has its own health impact and WHO recommends limiting the consumption of saturated and trans-fats (hydrogenated fats), sugars and salt in the diet, which are often found in snacks, processed foods and drinks [35]. Healthy diets should provide very low intake of TFAs (i.e.,
problems of fried products like potato chips, bonbollo and sanbusa. From the survey result the quality of oil used for frying 76.18% (32) respondents were not ask the quality of the oil used before frying. A change in IV were observed after a few days of frying until the value were reached from 11.982, 14.662, 19.975, 22.581 and 22.639 in sample 1 to 6 respectively. The maximum level suggested by FAO of United Nation (2013) is up to 10 milliequivalents of active oxygen/kg oil in refined oil. The increase in peroxide value was also suggested that the peroxide form during the storage condition. Furthermore, the lipid hydperoxe radical is formed when the heated oil is kept at room temperature and reused again in frying [34]. Freshly refined oils regularly have a PV, lower than 1 meq/kg oil and oil is considered to be rancid at a peroxide value above 10 meq/kg oil. [37] reported that a good quality frying vegetable oil should have a PV of less than 2 meq/kg. Studied done in-use frying oils collected during frying operations showed that PV ranges from 3.3 to 48.1 meq/kg. Then, in discarded samples the PV ranged from 4.7 to 247.5 meq/kg [38]. This can prove that a large proportion of the frying oils used in these commercial establishments were highly oxidized [35].

4.2.2. Free Fatty Acid (FFA)
Free fatty acid (FFA) is widely used by many food processors as an indicator of oil deterioration. This is because the method is relatively fast and reliable for monitoring the acidity of oil during frying [39]. FFA represents the percentage of fatty acids liberated from the triglyceride chain during heating due to hydrolysis and oxidation [40]. The formation of FFA is highly associated with the smoke point, with an increase in FFA content lowering the smoke point [36].

According to the result finding for FFA sample 1 to 6 before the oil used for frying the value were ranged from 0.9, 0.86, 0.64, 0.68, 0.47 and 0.77% respectively. The value of FFA in oil was kept increasing after it's used in frying until the value were reached from 2.182, 1.662, 1.961, 1.842, 2.281 and 2.132% in sample 1 to sample 6 respectively. Many researchers have reported an increase in FFA levels during frying which was due to the cleavage and oxidation of double bonds to form carbonyl compounds with the latter oxidizing to form low molecular fatty acids during frying experiments. The maximum allowed value of FFA varies depending on the type of food being fried. For the industrial production of potato chips, an FFA of 0.3% has been reported as the threshold for discarding used frying oil, whereas a maximum value of 1% FFA is usually allowed by the processors of pre-fried French fries (Ismail, 2005). Some countries even have regulation on the FFA threshold to discard frying oil: for example, Austria (1.25%), Belgium (2.5%), Germany (1%), Japan (1.25%) and the Netherlands (2.25%).

4.2.3. Iodine Value
The iodine value is a measure of unsaturation. A decrease in the iodine value is consistent with a decrease in the number of double bonds in heated oil as it becomes oxidized. The IV of oil sample from 1 to 6 used for frying purpose were ranged from 103, 94.6, 78.2, 62.91, 84.64 and 79.48 IN before frying. A change in IV were observed after a few days of frying in all oil’s samples (1 to 6) 92.3, 89.95, 71.48, 56.8, 73.67 and 69.99 IN respectively. The differences in iodine values of the oil during frying are also indicative of the increased rate of oxidation during frying. According to [37], a significant change in the iodine values can be observed when there is excessive deterioration of the oil.

4.2.4. Smoke Point
The smoke point of fat/oils are indicator of temperature at which a fat first starts to smoke. For most fats/oils, it is 216-232°C but it is lower for high lauric fat, emulsified fat and emulsifies. According to the finding the smoke point of repeated use of oils for frying purpose were decreased from 205, 221, 198, 177, 215 and 200°C for sample 1 to 6 respectively. These results agreed with Termizi and Siew, 2008 lowering the smoke point were observe during repeated use of oils and an increase of FFA.
5. Conclusions

The physicochemical properties of oil were investigated from sample selected randomly around wolkite town and these are free fatty acid, peroxide value, smoking point and iodine value of recycled oils. So, the present study shows that the oils used are not recommended as standards. The strong evidence suggestion and association of fried food consumption with a higher risk of developing chronic disease or at most infrequent to moderate fried food consumption within the context of an overall healthy dietary pattern. Generally the present study shows the recycled oil used were not as standards of FAO and also the people of fried food makers were not have awareness about the impacts of recycled oils.

References


