Physico-Chemical Analysis of Ground Water Quality of Bhanvad

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Abstract: Ground water samples were collected from diverse spaces of Bhanvad taluka of Jamnagar district (India) for investigation of their Physicochemical parameters. These Fifteen water samples from diverse spaces were analyzed for their physicochemical characteristics. Local community utilized this water for drinking and irrigation purpose. Laboratory investigations were carried out for analysis like, Temperature, Calcium-Magnesium hardness, pH, TDS, Chloride, Alkalinity, sulphate, phosphate and nitrate. These parameters are effectiveness in calculating quality of ground water. The key plan of our study is to find the quality of ground water in and around Bhanvad taluka and formulate it for drinking purposes after appropriate purification.

Keywords: Bhanvad Taluka, Physiochemical Parameters, Ground Water

1. Introduction

The wonder of the natural world is Water. In nature water is one of most vital requirement for all the living organisms. Therefore it is true that “No water No life”. For sustain of life under ground and surface water are essential natural resources, which is available in large quantity and it is a complimentary reward of nature [1-2]. In diverse forms and from diverse resources the water use for the human beings. Primarily two resources of drinking water, among them one is a surface water which includes River, lakes and second one is under ground water which are mainly from the escape of surface water and is detained in the subsoil and in prior rock. Ground water contains approximately 94% of total available water all over world. Major resource of drinking water in villages is under ground water which is available in the form of wells, bore wells or hand pumps [3-6].

As the population of world enlarge the require for cleans water also increases constantly. Public in numerous parts of the world are survival for lack of the fresh, drinkable water, therefore more protected water supplies are required.

Here we reported the physicochemical analysis of bore wells drinking water of Bhanvad territory. Bhanvad is situated in Jamnagar district of Gujarat. Bore wells water is usually utilize for Drinking and additional household functions in this area. The main sources of bore wells water pollution are the use of fertilizers and pesticides manure, lime, septic tank, refuse dump, etc. [7-9] For this reason, locality of this area required to utilize bore wells water for their drinking, domestic and other utilization. With the aim to evaluate water quality index, we have carried out the Physico-chemical analysis of bore wells drinking water.

2. Material and Methods

The Bore wells underground water samples of drinking water of Bhanvad taluka, were collected from 15 diverse
villages in the morning between 10 am to 12 am in plastic sampling bottles with essential protections. For the preparation of reagents and solutions double distilled water was utilized. The main water quality parameters measured for the inspection in this study are temperature, pH, dissolved oxygen (DO), total dissolved solid (TDS), total alkalinity, calcium and magnesium ions, sulphate, phosphate and nitrate contents [10]. Complexometric titration method was applied for estimation of Calcium and magnesium hardness of water [9, 11]. Volumetrically Chloride and Sulphate contents were determined [11]. The Various Physico-chemical parameters of ground water samples of Bhanvad taluka are shown in the Table-1.

<table>
<thead>
<tr>
<th>Sample Station</th>
<th>Temp °C</th>
<th>pH</th>
<th>TDS m/L</th>
<th>Cl mg/L</th>
<th>T.A. mg/L</th>
<th>C.H. mg/L</th>
<th>M.H. mg/L</th>
<th>SO$_4^{2-}$ mg/L</th>
<th>PO$_4^{3-}$ mg/L</th>
<th>NO$_3^{-}$ mg/L</th>
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<td>11.21</td>
<td>12.6</td>
<td>326.66</td>
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3. Results and Discussion

Due to diverse nature of soil contamination the results of the bore wells water samples varies [12]. Water temperature are generally influenced the entire metabolic, physiological activities and life of aquatic organisms.

Temperature

Temperature range was situating between 27.5°C to 33.5°C for present analysis.

$\text{pH} \\

The acidity, alkalinity and resulting value of the acidic basic interaction of a number of its mineral and organic components are depends on the pH value of drinking water. The corrosion in pipes starts when pH less than 6.5. Presences of Toxic metals in water increase the pH. The pH range is 4.0 mg/L for drinking water.

$\text{TDS (Total Dissolved Substance)} \\

TDS value should be below than 500 mg/L for drinking water according to WHO and Indian standards [12, 13]. TDS ranged from 205 mg/L to 1430 mg/L are found in the present study.

$\text{D.O. (Dissolved Oxygen)} \\

In the present analysis dissolved oxygen (D.O) found in between 5.8 mg/L to 8.4 mg/L and the minimum tolerance range is 4.0 mg/L for drinking water.

$\text{Chlorides} \\

The tolerance range for chloride is 200 to 1000 mg/L. In present study it found in between 37.2 mg/L to 145.65 mg/L.

$\text{Total Alkalinity} \\

We found total alkalinity range was from 342 mg/L to 666 mg/L.

$\text{Water hardness due to Calcium ion (Ca}^{2+}) \\

The calcium ion ranges from 7.97 mg/L to 5.12 mg/L with the tolerance range is 75 to 200 mg/L. Water provides total requirements Calcium in the body.

$\text{Water hardness due to Magnesium ion (Mg}^{2+}) \\

Magnesium ion ranges from 12.60 to 95.52 mg/L, their tolerance range is 50 to 100 mg/L.

$\text{Sulphate ion (SO}_4^{2-}) \\

Sulphate ion ranges from 40.79 mg/L to 347.99 mg/L/land their tolerance range is 200 to 400 mg/L. The diarrhea may induce due to high concentration of sulphate.

$\text{Phosphate ion (PO}_4^{3-}) \\

In the present study phosphate ranged from 3.55 mg/L to 15.05 mg/L, which is much higher than the prescribed values. It may be due to exploit of fertilizers and pesticides by the people of this area. Excess phosphate consumption could lead to the death of consumer.

$\text{Nitrate ion (NO}_3^{-}) \\

In the present study nitrate ranged from 82 mg/L to 362 mg/L and their tolerance range for 20mg/L to 45 mg/L, which are higher than the prescribed values it may be due the excess use of fertilizers and pesticides.

4. Conclusion

In continuous of our previous work [14] in present research paper we describe the study of various Physico-chemical analysis of bore wells water like, temperature, pH, dissolved oxygen, total dissolved solids, chloride, total
alkalinity, calcium and magnesium hardness, sulphate, phosphate nitrate ions. TDS value should be less than 500 mg/L for drinking water. All the sample shows suitable ranged prescribed by WHO and Indian standards except than sample 11, which was found higher. It can affect living organisms.

References


