GC-MS Analysis of Phyto-components from the Leaves of Senna alata L

Omotoyinbo Oluwasegun Victor, Sanni Morakinyo David*

Department of Biochemistry, Federal University of Technology, Akure, Ondo State, Nigeria

Email address: moraksanni@yahoo.co.uk (D. M. Sanni)

To cite this article:

Abstract: The biochemical constituents of extracts obtained from the leaves of Senna alata obtained from Akure, Nigeria is being reported. The chloroform-methanol extracts was analysed by gas chromatography – mass spectrometry (GC-MS) techniques. The main constituents of the extracts were 6-Octadecenoic acid (24.99 %), 2, 3-Dihydroxypropyl-9-octadecenoate (20.86 %) and Octadecanoic acid (18.08 %).

Keywords: Senna alata, GC-MS, Ethnomedicine

1. Introduction

It has been assumed that “there is a plant for every need on every continent”. Remarkably, this statement appears to be true. Finding healing powers in plants is an ancient idea [1]. The World Health Organization (WHO) estimates that 80 % of the people living in developing countries almost exclusively use traditional medicine. Medicinal plants used in traditional medicine should therefore be studied for safety and efficacy. It has been estimated that 14-28 % of higher plant species are used medicinally, which are only 15 % of all angiosperms that have been investigated chemically. According to the database, 74 % of pharmacologically-active plant derived drugs have been discovered after following up on ethnomedical use of the plant [2]. Naturally molecules derived from plant extracts offer a particularly exciting avenue for further research [3].

Senna alata L, (syn. Cassia alata L. Roxb.) is an erect tropical, annual herb with leathery compounded leaves. It belongs to the Fabaceae family. It grows up to about 8m tall and can be found in diverse habitats. This perennial shrub has erect waxy yellow spikes that resemble fat candles before the individual blossoms open. The large leaves are bilaterally-symmetrical opposed and fold together at night. The fruit is a pod, while the seeds are small and square [4, 5].

Senna alata has been reported to have very high medicinal values like antimicrobial property particularly against fungal dermatophytes and traditionally being used in the treatment of skin infections in man [6, 7]. Leaf extract is a good antioxidant [7]. The juice of the fresh leaf of Senna alata is universally recognised as a remedy for parasitic skin diseases, and is used in many eruptive and pustular skin infections by simply rubbing the crushed leaves alone or mixed with lime juice or oil [8]. In Sierra Leone, the leaves in form of an infusion are used as a purgative, and a strong decoction is sometimes given as an abortifacent or during labour to hasten delivery, as the juice expressed from the fresh leaves is taken along with lime juice for worms [8].

With the numerous ethnomedicinal use of Senna alata leaf in different cultures and places, this study reveals the bioactive compounds present in Senna alata chloroform-methanol leaf extracts by GC-MS analysis.

2. Materials and Methods

2.1. Collection of Plant and Preparation of Extract

Senna alata plant leaves were obtained from Oba Ile area of Akure and confirmed by Herb sellers at Oja Oba market in Akure. Plant was later identified and authenticated at the Department of Crop Science, Federal University of Technology Akure, Ondo State Nigeria.

Shade dried leaves of Senna alata, were powdered and was extracted by the modified Bligh and Dyer procedure [9]. 5 g of powdered sample were shaken at room temperature with 18 ml methanol–chloroform (2:1, v/v) for 2 hours and filtered using cheese cloth. The residue obtained was further shaken with 18 ml of methanol–chloroform (2:1, v/v) and 2 ml of water. This was followed by another round of filtration while
the residue was also washed with 3 ml of methanol–
chloroform (2:1; v/v). The three filtrate obtained were
combined in a separatory funnel with 5 ml of chloroform and
6 ml of water added and shaken before being allowed to
separate into different phases. The chloroform layer was
separated, diluted with benzene and concentrated. The
residual lipids were immediately dissolved in 0.1 ml
chloroform–methanol (1:1) for analysis.

2.2. GC-MS Analysis and Identification of Components

The GC-MS analysis of Senna alata leaf oil extract was
performed using a GC-MS QP2010 PLUS Shimadzu, Japan.
Interpretation of mass spectrum of GC-MS was done using
the database of National Institute Standard and Technology
(NIST), having more than 62,000 patterns. The mass
spectrum of the unknown component was compared with the
spectrum of the known components stored in the NIST
library. The name, molecular weight and structure of the
components of the test materials were ascertained.

3. Results and Discussion

GC-MS chromatogram of the chloroform-methanol leaf oil
extract of Senna alata (Figure 1) showed 11 peaks indicating
the presence of eleven compounds. The chemical compounds
identified in the extract of the leaf of Senna alata are
presented in Table 1. The GC-MS analysis revealed that the
presence of Methylpentadecanoate (0.47 %), n-Hexadecanoic
acid (9.49 %), Methylloctadec-9-enoate (0.93 %), 6-
Octadecenoic acid (24.99 %), Octadecanoic acid (18.08 %),
Glycerol-1,3-dipalmitate (2.99 %), 9-Octadecenoyl chloride
(1.19 %), 9-Octadecenal (5.49 %), 3-Hydroxypropyl-9-
Octadecenoate (3.64 %), 4-Dimethylsilyloxypentadecane
(11.87 %), 2,3-Dihydroxypropyl-9-octadecenoate (20.86 %).
The GC-MS analyses revealed that the lipid extract is
mainly composed of esters and long chain fatty acids.
Hexadecanoic acid has earlier been reported as the major
component in leaf extract of Kigelia pinnata [10] and
identified 17 compounds with n-Hexadecanoic acid and
Octadecanoic acid as the major compounds in the leaves of
Cleistanthus collinus. Esters and long chain fatty acids are
constituents of plant resin and essential oils extracted from
plants. Essential oils are volatile, natural, complex
compounds characterized by a strong odour and are formed
by aromatic plants as secondary metabolites [13]. In nature,
essential oils play an important role in the protection of
plants as antibacterial, antivirals, antifungals, insecticides and
also against herbivores by reducing their appetite for such
plants. They may also attract some insects, thereby favouring
the dispersion of pollons and seeds, or repel other undesirable
insects [13]. Medical professionals are more interested in the
medicinal properties of essential oils. Many oils show
antibacterial, fungicidal, relaxant, stimulating, antidepressant
effect and can be very effective therapeutic agent. Essential
oils are known for their therapeutical properties hence, used in
the treatment of various infections caused by both pathogenic
and non-pathogenic diseases. Pathogenic diseases caused by
bacterial, virus, and the fungi can be treated with essential
oils [14]. Therefore the phytochemicals reported in this
extract will account for the various pharmacological actions
these plants part possess. So far, there is no study that can
give a clear idea and be accurate on the mode of action of the
essential oils. Given the complexity of their chemical
composition, it is very likely that each of the constituents of
the essential oils as obtained from Senna alata leaf has its
own mechanism of action. Because of the variability of
amounts and profiles of the components of essential oils, it is
likely that their various medicinal activities is not due to a
single mechanism, but to several sites of action at the cellular
level and involving different modes of action. Thus, this type
of GC – MS analysis is the first step towards understanding
the nature of active principles in this medicinal plant and this
type of study will be helpful for further detailed study in line
with the biochemical and phytochemical functions mentioned
above.

Figure 1. GC-MS Profile of chloroform-methanol leaves extract of Senna alata.
Table 1. Bio-activity of phyto-components identified in the chloroform-methanol extracts of the leaf of Senna alata.

<table>
<thead>
<tr>
<th>No.</th>
<th>RT</th>
<th>Name of the compound</th>
<th>Molecular Formula</th>
<th>Molecular Weight</th>
<th>Peak Area %</th>
<th>Structure</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.</td>
<td>15.41</td>
<td>Methylpentadecanoate</td>
<td>C_{16}H_{32}O_{2}</td>
<td>256</td>
<td>0.47</td>
<td><img src="image1" alt="Structure" /></td>
</tr>
<tr>
<td>2.</td>
<td>15.82</td>
<td>n-Hexadecanoic acid</td>
<td>C_{16}H_{32}O_{2}</td>
<td>256</td>
<td>9.49</td>
<td><img src="image2" alt="Structure" /></td>
</tr>
<tr>
<td>3.</td>
<td>17.08</td>
<td>Methylloctadec-9-enoate</td>
<td>C_{18}H_{36}O_{2}</td>
<td>296</td>
<td>0.93</td>
<td><img src="image3" alt="Structure" /></td>
</tr>
<tr>
<td>4.</td>
<td>17.48</td>
<td>6-Octadecenoic acid</td>
<td>C_{18}H_{34}O_{2}</td>
<td>282</td>
<td>24.99</td>
<td><img src="image4" alt="Structure" /></td>
</tr>
<tr>
<td>5.</td>
<td>17.69</td>
<td>Octadecanoic acid</td>
<td>C_{18}H_{36}O_{2}</td>
<td>284</td>
<td>18.08</td>
<td><img src="image5" alt="Structure" /></td>
</tr>
<tr>
<td>6.</td>
<td>18.73</td>
<td>Glycerol-1,3-dipalmitate</td>
<td>C_{35}H_{68}O_{5}</td>
<td>568</td>
<td>2.99</td>
<td><img src="image6" alt="Structure" /></td>
</tr>
<tr>
<td>7.</td>
<td>19.79</td>
<td>9-Octadeconoyl chloride</td>
<td>C_{18}H_{34}ClO</td>
<td>300</td>
<td>1.19</td>
<td><img src="image7" alt="Structure" /></td>
</tr>
<tr>
<td>8.</td>
<td>20.19</td>
<td>9-Octadecenal</td>
<td>C_{18}H_{36}O</td>
<td>266</td>
<td>5.49</td>
<td><img src="image8" alt="Structure" /></td>
</tr>
<tr>
<td>9.</td>
<td>20.68</td>
<td>3-Hydroxypropyl-9-octadecanoate</td>
<td>C_{21}H_{40}O_{3}</td>
<td>340</td>
<td>3.64</td>
<td><img src="image9" alt="Structure" /></td>
</tr>
<tr>
<td>10.</td>
<td>21.64</td>
<td>4-Dimethylsilyloxypentadecane</td>
<td>C_{17}H_{38}O_{Si}</td>
<td>286</td>
<td>11.87</td>
<td><img src="image10" alt="Structure" /></td>
</tr>
<tr>
<td>11.</td>
<td>21.92</td>
<td>2,3-Dihydroxypropyl-9-octadecanoate</td>
<td>C_{21}H_{40}O_{4}</td>
<td>356</td>
<td>20.86</td>
<td><img src="image11" alt="Structure" /></td>
</tr>
</tbody>
</table>
References


